CLASS XII CHEMISTRY CBT NOVEMBER 2023 QP AND ANALYTICS

Q NO	CHAPTER	QUESTION	OPTIONS PROVIDED	CORRECT OPTION	EXPLAINATION	% ATTEMPTED CORRECTLY
01	CARBOXYLIC ACID (FUNCTIONAL GROUP III)	The natural bacterial fermentation of milk produces a compound called propanoic acid. Propanoic acid is partly responsible for the flavour of Swiss cheese. Only Propanoic acid Which starting materials could be used to produce propanoic acid? 1 CH ₃ CH ₂ CN 2 CH ₃ CH ₂ CHO 3 CH ₃ CH ₂ CHO	A. 1 & 2 B. 2 & 3 C. 1 & 3 D. 1,2 & 3	D	All the three compounds can be used for preparation of Final Product following Hydrolysis or Oxidation Mechanism	42.60 %
02		Alkaline Hydrolysis is also Known as	A. Saponification B. Esterification C. Hydration D. Alkalisation	А	Saponification is the Alkaline Hydrolysis of SoapsA	62.50 %
03		$CH_3COCI \xrightarrow{Pd, H_2} X$; then X is?	A. Ethanal B. Propanal C. Acetone D. Acetic Acid	A	Reduction of Aldehydes called Rosenmund Reaction	57.90 %
04		Assertion : $HCOOH$ formic acid add react with H_2SO_4 to form CO . Reason : H_2SO_4 is mild (moderate) oxidizing agent.	Assertion and Reason both are true but Reason is not the correct explanation of Assertion.		38 %	
05		Assertion: Carboxylic acids have higher boiling points than alkanes. Reason: Carboxylic acids are resonance hybrids.	Assertion and Reason both are true but Reason is not the correct explanation of Assertion.			55.40 %

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06	NITROGEN CONTAINING COMPOUNDS – AMMINES	Assertion: Aromatic 1°amines can be prepared by Gabriel phthalimide synthesis. Reason: Aryl halides undergo nucleophilic substitution with anion formed by phthalimide	If both Reason and Assertion are correct and Reason is correct Explanation	A	NA	45.30 %
07		Assertion: Only a small amount of HCl is required in the reduction of nitro compounds with iron scrap and HCl in the presence of steam. Reason: FeCl2 formed gets hydrolysed to release HCl during the reaction.	If both Reason and Assertion are incorrect	D	NA	02 %
08		Which test can tell the difference between p-chloroaniline and anilinium hydrochloride?	 A. Sandmeyer Test B. Victor Meyer Test C. Silver Nitrate Test D. Sodium Bi Carbonate Solution 	D	NA	08.50 %
09		Which of the following statements concerning methylamine is correct?	 A. Methylamine is stronger base than NH3 B. Methylamine is less basic thanNH3 C. Methylamine is slightly acidic D. Methylamine forms salts withalkali 	А		73.30 %
10		Which of these substances has a lower melting point than amine?	A. Alcohol B. Ether C. Carboxylic Acid D. Phosphine	D	Absence of Inter Molecular H- Bonding	48.90 %